## $A Q A B$

Please write clearly in block capitals.

Centre number $\square$ Candidate number


Surname
Forename(s)
Candidate signature

## GCSE

COMBINED SCIENCE: TRILOGY


Higher Tier
Physics Paper 1H
Wednesday 23 May 2018 Afternoon Time allowed: 1 hour 15 minutes

## Materials

For this paper you must have:

- a ruler
- a scientific calculator
- the Physics Equations Sheet (enclosed).


## Instructions

- Use black ink or black ball-point pen.
- Fill in the boxes at the top of this page.
- Answer all questions in the spaces provided.
- Do all rough work in this book. Cross through any work you do not want to be marked.
- In all calculations, show clearly how you work out your answer.

| For Examiner's Use |  |
| :---: | :---: |
| Question | Mark |
| 1 |  |
| 2 |  |
| 3 |  |
| 4 |  |
| 5 |  |
| 6 |  |
| TOTAL |  |

## Information

- The maximum mark for this paper is 70 .
- The marks for questions are shown in brackets.
- You are expected to use a calculator where appropriate.
- You are reminded of the need for good English and clear presentation in your answers.

| 0 | 1 | Figure 1 shows two models of the atom. |
| :--- | :--- | :--- |

Figure 1


Plum pudding model


Nuclear model

| $\mathbf{0}$ | $\mathbf{1}$ | $\mathbf{1}$ |
| :--- | :--- | :--- |

Choose the answers from the box.

| atom | electron | nucleus |
| :---: | :---: | :---: |
| neutron | orbit | proton |


| $\mathbf{0}$ | $\mathbf{1}$ | $\mathbf{2}$ Explain why the total positive charge in every atom of an element is always the same. |
| :--- | :--- | :--- | [2 marks]

$\qquad$
$\qquad$
$\qquad$
$\qquad$

Alpha particles were fired at a thin film of gold at a speed of $7 \%$ of the speed of light.
Determine the speed of the alpha particles.
Speed of light $=300000000 \mathrm{~m} / \mathrm{s}$
$\qquad$
$\qquad$
Speed $=$ $\qquad$ $\mathrm{m} / \mathrm{s}$

| 0 | 1 | 4 | Figure 2 shows two atoms represented as solid spheres. |
| :--- | :--- | :--- | :--- |

Figure 2


Hydrogen


A hydrogen atom has a radius of $2.5 \times 10^{-11} \mathrm{~m}$
Determine the radius of a magnesium atom.
Use measurements from Figure 2
$\qquad$
$\qquad$
Radius $=$ $\qquad$ m

02 A student wanted to determine the density of the irregular shaped object shown in Figure 3

Figure 3


| $\mathbf{0}$ | $\mathbf{2}$ | $\mathbf{1}$ Plan an experiment that would allow the student to determine the density |
| :--- | :--- | :--- | of the object.

$\qquad$
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| $\mathbf{0}$ | $\mathbf{2} .2$ | $\mathbf{2}$ Another student did a similar experiment. |
| :--- | :--- | :--- |

He determined the density of five common plastic materials.
Table 1 shows the results.
Table 1

| Plastic material | Density in $\mathbf{~ k g} / \mathbf{m}^{3}$ |
| :--- | :---: |
| Acrylic | 1200 |
| Nylon | 1000 |
| Polyester | 1380 |
| Polystyrene | 1040 |
| PVC | 1100 |

Figure 4 shows the results plotted in a bar chart.
Figure 4


## Complete Figure 4

You should:

- Write the correct scale on the y-axis.
- Draw the bars for polyester, polystyrene and PVC.

The student determined the density of the material three times.
Table 2 shows the results.


## Table 2

|  | Density in $\mathbf{~ k g} / \mathbf{m}^{3}$ |
| :--- | :---: |
| 1 | 960 |
| 2 | 1120 |
| 3 | 1040 |

Determine the uncertainty in the student's results.
$\qquad$
$\qquad$
Uncertainty = $\qquad$ $\mathrm{kg} / \mathrm{m}^{3}$

| 0 | 3 | Figure 5 shows a diver. |
| :--- | :--- | :--- |

The diver is using a canister of compressed air so that he can breathe underwater.
Figure 5


| $\mathbf{0}$ | $\mathbf{3}$. | $\mathbf{1}$ Which two sentences describe the movement of the air particles in the canister? |
| :--- | :--- | :--- |

[2 marks]
Tick two boxes.

They vibrate about a fixed position.

They move in random directions.



The motion of all the particles is predictable.


They move with a range of different speeds.


They move in circular paths.


| $\mathbf{0}$ | $\mathbf{3}$ | $\mathbf{2}$ The temperature of the air inside the canister increases. |
| :--- | :--- | :--- | :--- |

What happens to the movement of the air particles?
$\qquad$

| $\mathbf{0}$ | $\mathbf{3}$. | $\mathbf{3}$ It could be dangerous if the temperature of the air inside the canister increased by a |
| :--- | :--- | :--- | :--- | large amount.

Explain why.
$\qquad$
$\qquad$
$\qquad$
$\qquad$

A canister of air was tested to find out how the pressure changed when it was used by a diver.

- Air was allowed to escape from the canister.
- The pressure of the air in the canister was recorded every 5 minutes for 80 minutes.

Figure 6 shows the results.
Figure 6


| 0 | 3 | 4 |
| :--- | :--- | :--- |

## Use Figure 6

Atmospheric pressure $=$ $\qquad$ MPa

| 0 | 3 | 5 |
| :--- | :--- | :--- |
| Divers can safely stay underwater until the pressure of the air in the canister has |  |  | reduced to $25 \%$ of its original value.

Determine the maximum time the diver can safely stay underwater.
Use Figure 6
$\qquad$
$\qquad$
Time $=$ $\qquad$ minutes

| $\mathbf{0}$ | $\mathbf{3}$. | 6 |
| :--- | :--- | :--- | What happens to the volume of the air when it is released from the canister?

## Turn over for the next question

| $\mathbf{0}$ | $\mathbf{4} \quad$ The Chernobyl disaster was a nuclear accident that happened in 1986 |
| :--- | :--- |

Radioactive isotopes were released into the environment.
The radioactive isotopes emitted alpha, beta and gamma radiation.

| 0 | 4 | 1 |
| :--- | :--- | :--- | What is an alpha particle?

Tick one box.

2 charged particles and 2 neutral particles.

2 charged particles and 4 neutral particles. $\square$

4 charged particles and 2 neutral particles.


4 charged particles and 4 neutral particles.


| $\mathbf{0}$ | $\mathbf{4}$ |
| :--- | :--- | :--- |
| $\mathbf{2}$ Which statement about beta radiation is true? |  |

Tick one box.

It is the fastest moving type of radiation.

It is the type of radiation with a negative charge.

It is the type of radiation with the greatest mass.


It is the type of radiation with the greatest range in air.


| 0 | $\mathbf{4}$ | $\mathbf{3}$ Which statement about gamma radiation is true? |
| :--- | :--- | :--- |

Tick one box.

It is a low frequency electromagnetic wave.

It causes the charge of the nucleus to change.

$\square$

It causes the mass of the nucleus to change. $\square$
It has a very long range in air. $\square$

## Question 4 continues on the next page

| Table 3 shows the half-lives of two of the radioactive isotopes that |
| :--- |
| environment. |
| Table 3 |


| Isotope | Half-life |
| :--- | :---: |
| Caesium-137 | 30 years |
| lodine-131 | 8 days |


| $\mathbf{0}$ | $\mathbf{4}$ | $\mathbf{4}$ | A soil sample was taken from the area around Chernobyl in 1986 |
| :--- | :--- | :--- | :--- |

The soil sample was contaminated with equal amounts of caesium-137 and iodine-131

Explain how the risk linked to each isotope has changed between 1986 and 2018 Both isotopes emit the same type of radiation.
$\qquad$
$\qquad$
$\qquad$
$\qquad$
$\qquad$
$\qquad$
$\qquad$

| 0 | 4 | 5 |
| :--- | :--- | :--- |
| 5 |  |  | $1 / 32$ of its original value.

$\qquad$
$\qquad$
$\qquad$
Year $=$ $\qquad$

| 0 | 5 | Figure 7 shows cavity wall insulation being installed in the wall of a house. |
| :--- | :--- | :--- |

Figure 7


| $\mathbf{0}$ | $\mathbf{5}$. | $\mathbf{1}$ Explain how the wall reduces unwanted energy transfers. |
| :--- | :--- | :--- |

$\qquad$
$\qquad$
$\qquad$
$\qquad$
$\qquad$
$\qquad$

Question 5 continues on the next page

| $\mathbf{0}$ | $\mathbf{5}$. | $\mathbf{2}$ |
| :--- | :--- | :--- | The cavity insulation was tested.

- The heating inside the house was switched off.
- The temperature inside the house was measured every 20 minutes for 2 hours.

Table 4 shows the results.
Table 4

| Time in minutes | Temperature in ${ }^{\circ} \mathbf{C}$ |
| :--- | :---: |
| 0 | 25.0 |
| 20 | 20.8 |
| 40 | 17.4 |
| 60 | 14.5 |
| 80 | 12.1 |
| 100 | 10.0 |
| 120 | 8.4 |

Determine the temperature inside the house after 30 minutes.
$\qquad$
$\qquad$
$\qquad$
$\qquad$
Temperature $=$ $\qquad$ ${ }^{\circ} \mathrm{C}$

| 0 | 5 | 3 |
| :--- | :--- | :--- |

Figure 8


Describe how different energy stores are changed by the boiler.
$\qquad$
$\qquad$
$\qquad$
$\qquad$

Calculate the power of the boiler.
Write any equation that you use.
$\qquad$
$\qquad$
$\qquad$
Power = $\qquad$ W

Turn over for the next question

| 0 | 6 |
| :--- | :--- | A student built a circuit using filament lamps.


| $\mathbf{0}$ | $\mathbf{6} .1$ | Sketch a current potential difference graph for a filament lamp on Figure 9 |
| :--- | :--- | :--- |

Figure 9


Potential difference

Figure 10 shows the circuit with two identical filament lamps.
Figure 10


| 0 | 6. | 2 |
| :--- | :--- | :--- |
| Compare the currents $I_{1}, I_{2}$ and $I_{3}$ |  |  |

$\qquad$
$\qquad$
$\qquad$

Question 6 continues on the next page

| $\mathbf{0}$ | $\mathbf{6}$ | $\mathbf{3}$ Calculate the charge that flows through the cell in 1 minute. |
| :--- | :--- | :--- |

Each filament lamp has a power of 3 W and a resistance of $12 \Omega$
Write any equations that you use.
Give the unit.
$\qquad$
$\qquad$
$\qquad$
$\qquad$
$\qquad$
$\qquad$
Charge $=$ $\qquad$
Unit $=$ $\qquad$

| $\mathbf{0}$ | $\mathbf{6}$. | $\mathbf{4}$ | The student builds a different circuit. |
| :--- | :--- | :--- | :--- |

Figure 11 shows the circuit.
Figure 11


Explain how the readings on both meters change when the environmental conditions change.
$\qquad$
$\qquad$
$\qquad$
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$\qquad$

END OF QUESTIONS

## There are no questions printed on this page

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