AQA, OCR, Edexcel

## **GCSE Science**

## **GCSE Chemistry**

Materials Answers



Total Marks: /30

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Q1: Give the four examples whereby life cycle assessments (LCAs) are carried out to assess the environmental impacts of certain products.

- 1. Extracting and processing raw materials (1 mark)
- 2. Manufacturing and packaging (1 mark)
- 3. Use and operation during its lifetime (1 mark)
- 4. Disposal at the end of its useful life, including transport and distribution at each stage (1 mark)

(4 marks)

Q2: Describe some of the problems with LCAs.

A= Allocating numerical values to pollutant effects is not straightforward and requires value judgements (1 mark) LCA is not purely objective/ It is subjective/ based on opinion (1 mark)

(2 marks)

## Ways of reducing the use of resources

Q4: Metals, glass, building materials, clay ceramics and most plastics are produced from limited raw materials. Explain how metals and glass can be recycled.

Metals

A= metals can be recycled by melting and recasting or reforming into different products (1 mark)

Glass

A= Glass bottles can be crushed and melted (1 mark) to make different glass products

(2 marks)

Q5: What is corrosion?

A= corrosion is the destruction of materials by chemical reactions with substances in the environment.

(1 mark)

Q6: Give an example of corrosion.

A= Rusting

(1 mark)

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Q7: Give an example of how corrosion can be prevented.
A = Applying a coating that act as a barrier
(1 mark)
Q8:
Bronze is a metal alloy. What is it made up of?
A= copper and tin (2 marks)
Q9:
Brass is also a metal alloy. What is it made up of?
A= copper and zinc
(2 marks)
Q10: Gold alloy is used for jewellery along with which other metals?
A = Silver, copper and zinc
(3 marks)
Q11: The proportion of gold in the alloy is measured in carats. Match the carats with the percentage of gold.
24 carat → <u>100</u> % gold 18 carat → <b>75</b> % gold
(2 marks)
Q12: Complete the gaps in these sentences.
Steels are alloys of iron (1 mark)_that contain specific amounts of carbon (1 mark) and other metals.
High <b>carbon (1 mark)</b> steel is strong but <b>brittle (1 mark)</b> . Low <b>carbon (1 mark)</b> steel is softer and more easily <b>shaped (1 mark)</b> .
Steels containing <b>chromium (1 mark)</b> and <b>nickel</b> (stainless steels) are hard and resistant to <b>corrosion</b> (1 mark).
(9 marks)

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