## Science test

## TIER

## Paper 2

First name $\qquad$

Last name

School

## Remember

- The test is 1 hour long.
- You will need: pen, pencil, rubber, ruler, protractor and calculator.
- The test starts with easier questions.
- Try to answer all of the questions.
- The number of marks available for each question is given below the mark boxes in the margin. You should not write in this margin.
- If you are asked to plan an investigation, there will be space for you to write down your thoughts and ideas.
- Do not use any rough paper.
- Check your work carefully.
- Ask your teacher if you are not sure what to do.

1. (a) The diagrams below show the patterns produced on an oscilloscope by three different sound waves.


A


B


C
(i) Which two waves have the same loudness?

Write the letters.
$\qquad$ and $\qquad$
How do the diagrams show this?
$\qquad$
$\qquad$
(ii) Which two waves have the same pitch?

Write the letters.
$\qquad$ and $\qquad$
How do the diagrams show this?
$\qquad$
$\qquad$
(iii) Shuli is listening to a sound that produces the pattern below.


Describe how the sound that Shuli hears changes between X and Y .
(b) The table below shows the maximum time a person can listen to music at different sound levels without damage to the ear.

| sound level (decibels) | maximum time (hours) |
| :---: | :---: |
| 86 | 8 |
| 88 | 4 |
| 90 | 2 |
| 92 | 1 |
| 94 | 0.5 |

Estimate the maximum time a person could listen to a sound of 87 decibels.
$\qquad$ hours
(c) The diagram below shows part of the human ear.


What happens to the ear drum as a sound gets louder?
$\qquad$
$\qquad$ $-$
2. Jenny put a spring over a wooden rod.

She pressed the spring down 2 cm .

She let go of the spring and measured the height it reached.

not to scale

Jenny repeated her experiment. She pressed the spring down more each time. Her results are shown in the graph below.

(a) Use Jenny's graph to complete the table below.

| distance the spring was <br> pressed down (cm) | height the spring <br> reached (cm) |
| :---: | :---: |
| 2 |  |
| 3 |  |
| 4 |  |

(b) Jenny said, 'If I double the distance I press the spring down, the height it reaches will also double'.

How do the results show she was wrong?
$\qquad$
$\qquad$
(c) This diagram shows the moving spring in three different positions.

Complete the sentences below by choosing words from the box.
You can use each word more than once.

| most | some | least |
| :--- | :--- | :--- |


(i) When the spring is moving at $\mathbf{B}$ it has $\qquad$ kinetic energy and
$\qquad$ gravitational potential energy.
(ii) When the spring reaches $\mathbf{C}$ it has $\qquad$ gravitational potential energy and $\qquad$ kinetic energy.
(iii) When the spring stops at $\mathbf{A}$ it has $\qquad$ kinetic energy and
$\qquad$ gravitational potential energy.
3. (a) The diagram below shows a fish tank.

The surface of the water acts like a mirror.
The fish can see the snail reflected in the surface of the water.


Draw a ray of light which passes from the snail, and reflects from the surface, to show how the fish can see the snail. Use a ruler.

Put arrows on the ray of light.
(b) Andrew is looking at the snail.


When a ray of light passes from water to air it changes direction.
(i) Draw a ray of light from the snail to Andrew to show how Andrew can see the snail. Use a ruler.

Put arrows on the ray of light.


1 mark

(ii) What is the name given to this change in the direction of a ray of light?
4. Paul had four substances:

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citric acid
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He dissolved 1 g of each substance in $20 \mathrm{~cm}^{3}$ of distilled water. He used universal indicator to find the pH of each solution.
(a) (i) Sugar solution does not change the colour of green universal indicator.

What does this tell you about sugar solution?
Tick the correct box.


It is neutral.


It is sweet.

(ii) Suggest the pH of citric acid.
(iii) Indigestion tablets neutralise acid in the stomach.

What does this tell you about indigestion tablets?
(b) Complete the flow chart below with the names of the substances in the boxes.

| citric acid copper sulphate $\quad$ indigestion tablet $\quad$ sugar |
| :--- | :--- |


5. The drawing below shows a gemstone set in a gold ring.


Crystals of gemstones are found in different rocks.
(a) There are three groups of rocks:

| igneous metamorphic | sedimentary |
| :--- | :--- | :--- |

(i) Crystals can be found in rocks that have been changed into different rocks by high temperature and high pressure.

Which group of rocks is formed in this way?
(ii) Crystals can be found in rocks formed by the cooling of hot magma.

Which group of rocks is formed in this way?
(b) How does the rate at which magma cools affect the size of the crystals formed?
$\qquad$
$\qquad$
(c) Gemstones called rubies are made from an aluminium compound with the formula $\mathrm{Al}_{2} \mathrm{O}_{3}$.

The chemical symbol for aluminium is Al.
(i) Give the name of the element that is combined with aluminium in this compound.
$\qquad$
(ii) Suggest the name of the compound with the formula $\mathrm{Al}_{2} \mathrm{O}_{3}$.
$\qquad$
(iii) How many atoms are there in the formula $\mathrm{Al}_{2} \mathrm{O}_{3}$ ?
$\qquad$
(d) (i) The gemstone in the drawing is set into a gold ring.

Gold is an element that is found in rocks.
Gold is never found combined with other elements.
Part of the reactivity series of metals is shown below.

| more reactive | aluminium <br> zinc |
| :---: | :--- |
| less reactive | lead <br> copper |

Where should gold be placed in this reactivity series?
(ii) The more reactive metals react with acids.

Complete the word equation for the reaction of zinc with hydrochloric acid.

$$
\begin{gathered}
\text { zinc }+ \text { hydrochloric } \longrightarrow \\
\text { acid }
\end{gathered}
$$ $+$ $\qquad$



6. The table below shows the mass of six nutrients in $100 \mathrm{~cm}^{3}$ of three types of milk.

| nutrient | $100 \mathbf{c m}^{3}$ of <br> human milk | $100 \mathbf{c m}^{3}$ of <br> cows' milk | $100 \mathbf{~ c m}^{3}$ of <br> milk made from <br> baby-milk powder |
| :--- | :---: | :---: | :---: |
| carbohydrate (g) | 7.4 | 5.0 | 7.2 |
| fat (g) | 4.2 | 3.7 | 3.6 |
| protein (g) | 1.1 | 3.5 | 1.5 |
| calcium (mg) | 35.0 | 120.0 | 49.0 |
| iron $(\mathrm{mg})$ | 0.075 | 0.05 | 0.9 |
| vitamin $\mathrm{C}(\mathrm{mg})$ | 3.8 | 1.5 | 6.9 |

(a) A scientist compared the three types of milk.

Why was it a fair comparison?
$\qquad$


1 mark

(b) Both human milk and milk made from baby-milk powder contain more sugar than cows' milk.

Which data in the table supports this?
(c) Why do we need calcium in our diet?
(d) (i) Baby-milk powder is made from cows' milk.

What evidence is there in the table that iron is added when making baby-milk powder?
$\qquad$
$\qquad$
(ii) Why do we need iron in our diet?
$\qquad$
(e) A pupil said, 'There is more vitamin C than protein in human milk'.

How can you tell from the table that the pupil was wrong?
$\qquad$
$\qquad$
7. The diagram below shows a plant cell.

(a) In which part of a plant would you find this type of cell?


1 mark


1 mark


1 mark


1 mark
(c) Give the names of two labelled parts that are not present in animal cells.

1. $\qquad$
2. $\qquad$
(d) Tick one box in each row to show whether the statement is true for photosynthesis or for respiration.

| statement | photosynthesis | respiration |
| :--- | :--- | :--- |
| carbon dioxide is produced |  |  |
| light is needed |  |  |
| it occurs in plants and animals |  |  |
| oxygen is produced |  |  |

8. Mary used the apparatus below to test the strength of an electromagnet. She used the reading on the newton meter to measure the force of the magnet on the iron disc.

(a) Explain why the reading on the newton meter increases when a current passes through the coil.
$\qquad$
$\qquad$
$\qquad$
$\qquad$
(b) When a current passes through the coil, some of the electrical energy is changed to thermal energy.
What would happen to the coil if the current passing through it was too large?
(c) Mary made two electromagnets, one with 100 turns of wire in the coil and one with 200 turns.
She varied the current through the coil of each electromagnet.
She measured the force of each electromagnet on the iron disc.
The graph shows her results.


Write two conclusions that Mary could make from these results.

1. $\qquad$
$\qquad$
2. $\qquad$
$\qquad$

3. Neera and Tom dissolved different masses of salt in $500 \mathrm{~cm}^{3}$ of water. They measured the temperature at which each salt solution boiled.

(a) They wrote down the variables that might affect the investigation.
temperature of the
laboratory
mass of salt
dissolved in water

| starting temperature |
| :---: |
| of the water |

> boiling point of salt solution

## volume of water

## type of salt used

(i) What is the independent variable (the variable they changed) in their investigation?
$\qquad$
(ii) What is the dependent variable (the variable they measured) in their investigation?
$\qquad$
(iii) Which variable above would affect the experiment the least?
$\qquad$
(b) Neera and Tom plotted their results and drew the graphs shown below.


(i) How can you tell from the graphs that Neera and Tom started with pure water?
$\qquad$
$\qquad$

(ii) Why is Tom's line of best fit better than Neera's line of best fit?
$\qquad$
$\qquad$ _
10. Diagram A represents a gas in a container.

The gas can be compressed by moving the piston to the right.

(a) (i) How can you tell that the substance in the container is a gas?
$\qquad$
$\qquad$
(ii) How can you tell from the diagram that the gas is pure?
$\qquad$
$\qquad$
(b) The piston is moved to the right as shown in diagram B.


How can you tell, from diagram B, that the pressure of the gas has increased?
$\qquad$
$\qquad$
(c) Diagram $\mathbf{C}$ shows what happened to the molecules after the gas was compressed more.

(i) How can you tell that a chemical reaction happened when the gas was compressed?
$\qquad$
$\qquad$


1 mark
(ii) The mass of the gas in both diagrams $\mathbf{B}$ and $\mathbf{C}$ was 0.3 g .

Why did the mass of the gas not change when it was compressed?
$\qquad$
$\qquad$
(iii) Complete the table below with the correct chemical formula of each substance. Use the key to help you.

| substance | formula |
| :---: | :---: |
| 00 |  |
| 00 |  |
| 6 |  |


| Key |
| :---: |
| O nitrogen |
| Ooxygen |

(iv) What is the name of the substance represented by the symbol $\bigcirc$ ?
$\qquad$
11. The bar chart below shows how the number of cigarettes smoked is linked to the percentage of deaths from heart disease in the total male population.

(a) Use the information in the bar chart to write two conclusions about the relationship between smoking and the number of male deaths from heart disease.

1. $\qquad$
$\qquad$
2. $\qquad$
$\qquad$
(b) Smoking can cause fat to be deposited in the arteries to the heart muscle.

Explain how this could prevent the heart muscle from working properly.
$\qquad$
$\qquad$
$\qquad$
$\qquad$
$\qquad$
(c) The drawing below shows part of the lining of the airway leading into the lungs.

(i) Describe how mucus and cilia help to keep the airway free of dust and bacteria.
mucus $\qquad$
$\qquad$
cilia $\qquad$
(ii) Cigarette smoke contains tar.

What effect does tar have on the cilia?
$\qquad$
$\qquad$
12. A builder tried to remove a wooden post from the ground by pulling with a rope.

(a) (i) The builder attached a rope to hole $\mathrm{A}, 0.8 \mathrm{~m}$ above the ground. He pulled with a horizontal force of 300 N .

Calculate the turning moment about the pivot $P$.
Give the unit.
$\qquad$
$\qquad$
(ii) He then attached a rope to hole $B, 1.6 \mathrm{~m}$ above the ground. He pulled with a horizontal force.

What force would produce the same turning moment as before?
$\qquad$ N
(b) The post breaks off and falls on the ground as shown.

0


The weight of the broken post is 120 N .
The area in contact with the ground is $0.2 \mathrm{~m}^{2}$.
Calculate the pressure of the broken post on the ground.
Give the unit.

## END OF TEST

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