

## Density Mass Volume

Please write clearly in block capitals

Forename:

Surname:

### Materials

For this paper you must have:

- mathematical instruments



You **can** use a calculator.

### Instructions

- Use black ink or black ball-point pen. Draw diagrams in pencil.
- Fill in the boxes at the top of this page.
- Answer all questions.
- You must answer the questions in the spaces provided. Do not write outside the box around each page or on blank pages.
- Do all rough work in this book. Cross through any work you do not want to be marked.

### Information

- The marks for questions are shown in brackets.
- You may ask for graph paper, tracing paper and more answer paper. These must be tagged securely to this answer book.

### Advice

- In all calculations, show clearly how you work out your answer.

- 1 The table below shows the density, mass and volume of different objects.

Object	Mass	Volume	Density
A	27 kg	1500 cm <sup>3</sup>	
B		250 m <sup>3</sup>	96.2 g/m <sup>3</sup>
C	8.1 g		27 g/cm <sup>3</sup>

- 1(a) Calculate the density of object A in g/m<sup>3</sup>

[2 marks]

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Answer \_\_\_\_\_

- 1(b) Complete the table by filling in the empty spaces with values including units.

[3 marks]

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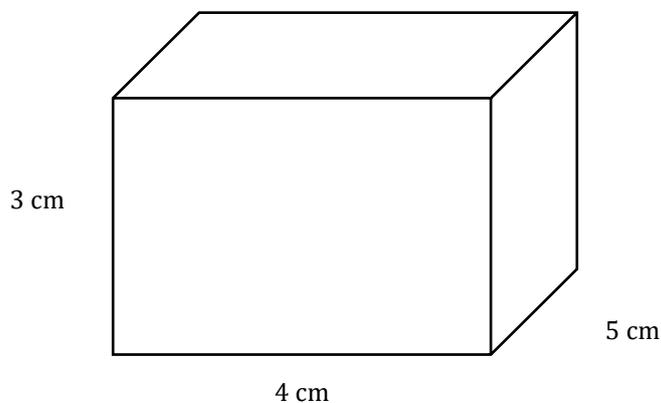
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Turn over for next question

Turn over ►

- 2 The diagram shows a wooden block with density  $0.57 \text{ g/cm}^3$



Not drawn accurately

Calculate the mass of the block.

[3 marks]

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Answer \_\_\_\_\_ g



### GCSE Maths Revision Guide

- ✓ GCSE Maths Course 9-1 Revision Guide
- ✓ Exam Questions Included
- ✓ All exam boards - AQA, OCR, Edexcel, WJEC
- ✓ Suitable for higher and foundation tiers

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Turn over ►

- 3(a)** Iron has a density of  $7.8 \text{ g/cm}^3$   
Calculate the mass of a  $3 \text{ cm}^3$  lump of iron.

[2 marks]

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Answer \_\_\_\_\_

- 3(b)** Aluminium has a density of  $2.7 \text{ g/cm}^3$   
Calculate the difference between the volume of a 5 g lump of iron and a 5 g lump of aluminium.

[3 marks]

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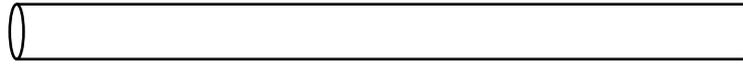
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Answer \_\_\_\_\_

- 4 A steel rod is in the shape of a cylinder, shown below.  
The steel rod has a density of  $9.8 \text{ g per cm}^3$ .  
The rod has a volume of  $60 \text{ cm}^3$ .

Not drawn  
accurately



Steel Rod

Calculate the mass of the rod in grams.

[2 marks]

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Answer \_\_\_\_\_



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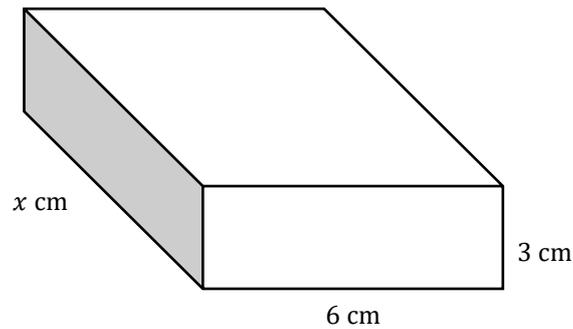
Turn over ►

5 The diagram below shows a cuboid.

Width is 6 cm

Height is 3 cm

Length is  $x$  cm



Not drawn  
accurately

5(a) The cuboid is made from wood and has a mass of 233.1 g.

The density of wood is  $1.85 \text{ g/cm}^3$ .

Calculate the volume of the cuboid.

[2 marks]

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Answer \_\_\_\_\_

5(b) Hence, or otherwise, find the missing length  $x$  of the cuboid.

[1 mark]

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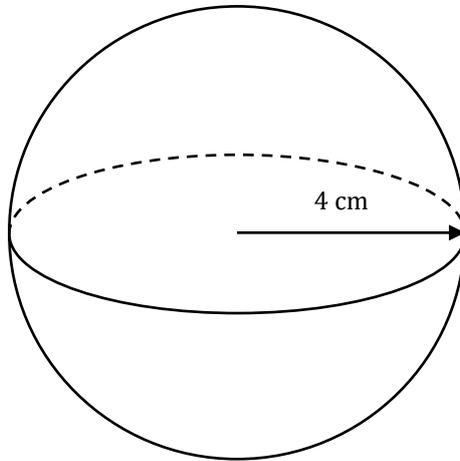
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Answer \_\_\_\_\_ cm

- 6 The diagram shows a spherical glass paperweight with a radius of 4 cm.



Not drawn  
accurately

The density of glass is  $8 \text{ g/cm}^3$ .

$$\text{Volume of a sphere} = \frac{4}{3}\pi r^3$$

Calculate the mass of the paperweight.

Give your answer correct to 3 significant figures.

**[3 marks]**

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Answer \_\_\_\_\_

**End of Questions**